

A potential habitable zone within the subsurface at the equatorial region on Mars

Alian Wang



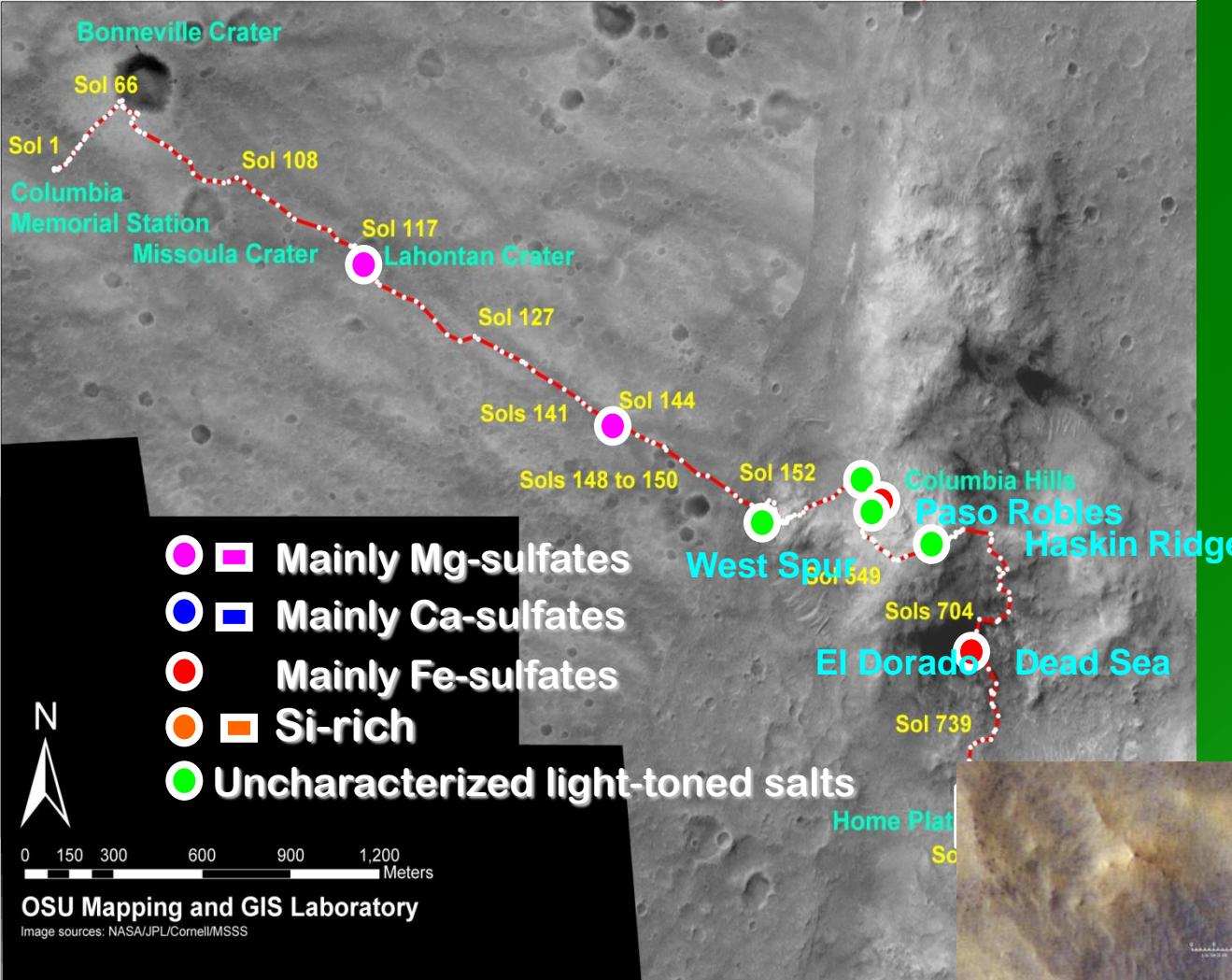
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Content

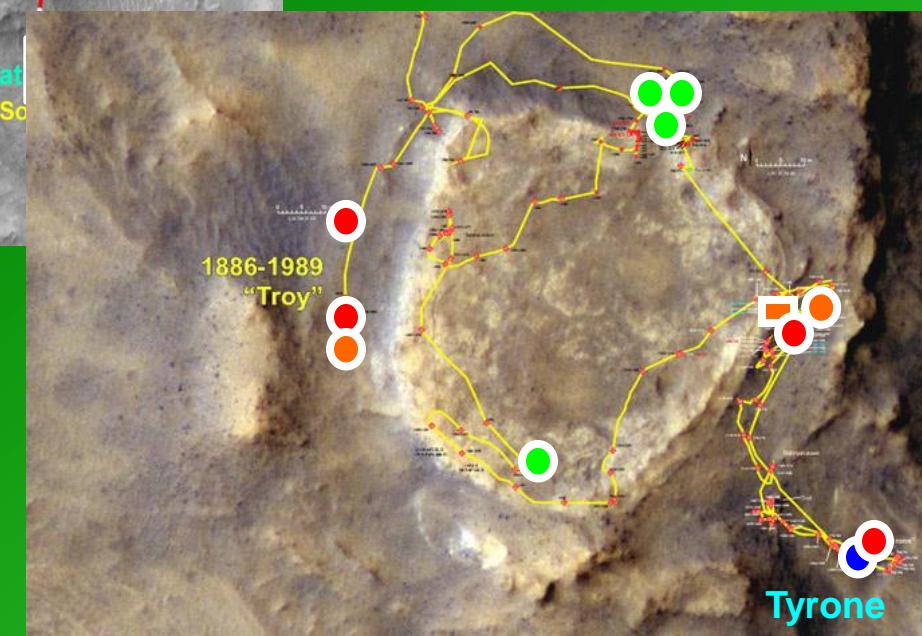
- Observations at Gusev on Mars;
- Observations at a terrestrial analog site;
- Extrapolation from a thermal model
(Mellon et al., 2004);
- Laboratory observations @ low T.

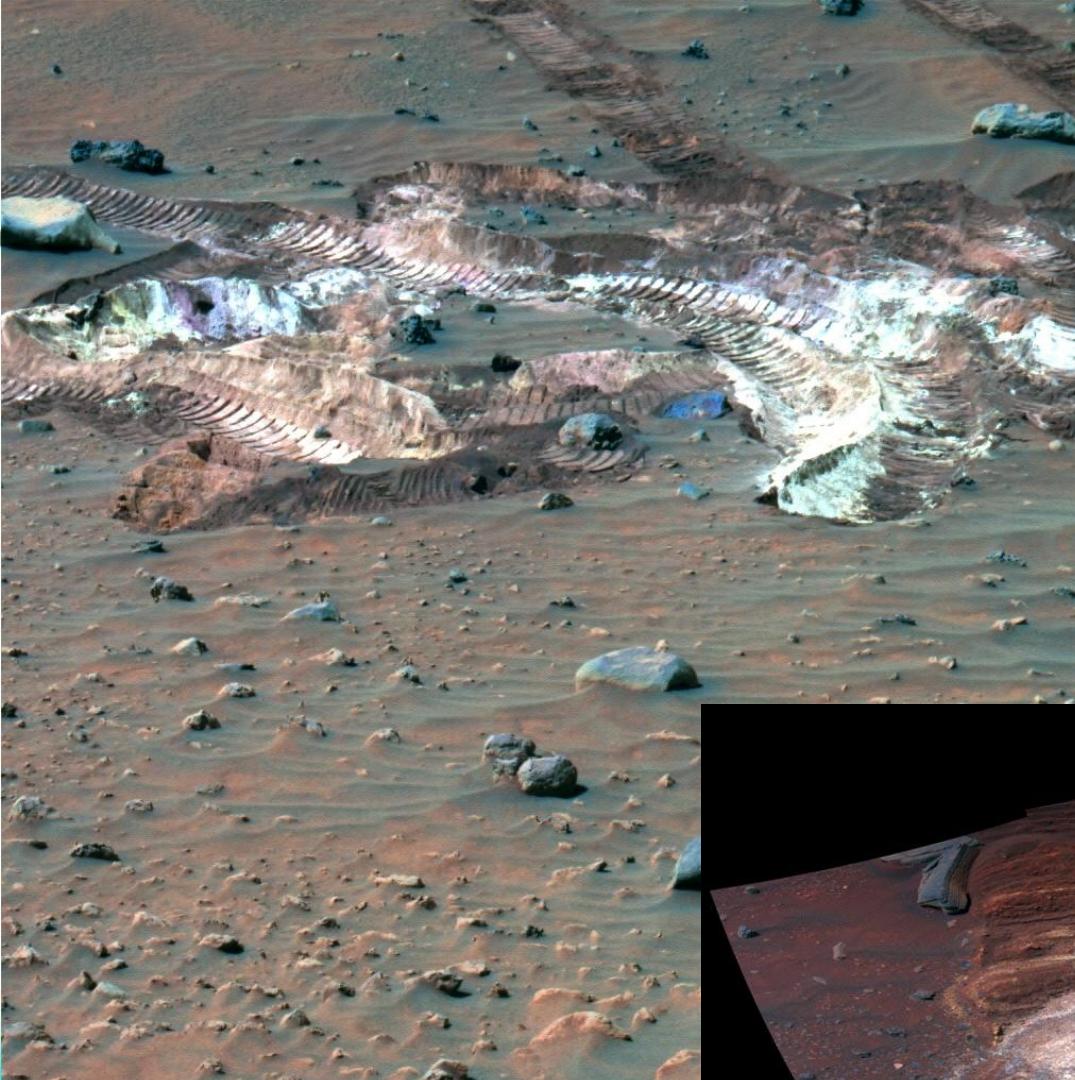
Mars salt-rich subsurface:

Low T, high RH, liquid H₂O film
→ habitable or habited ?



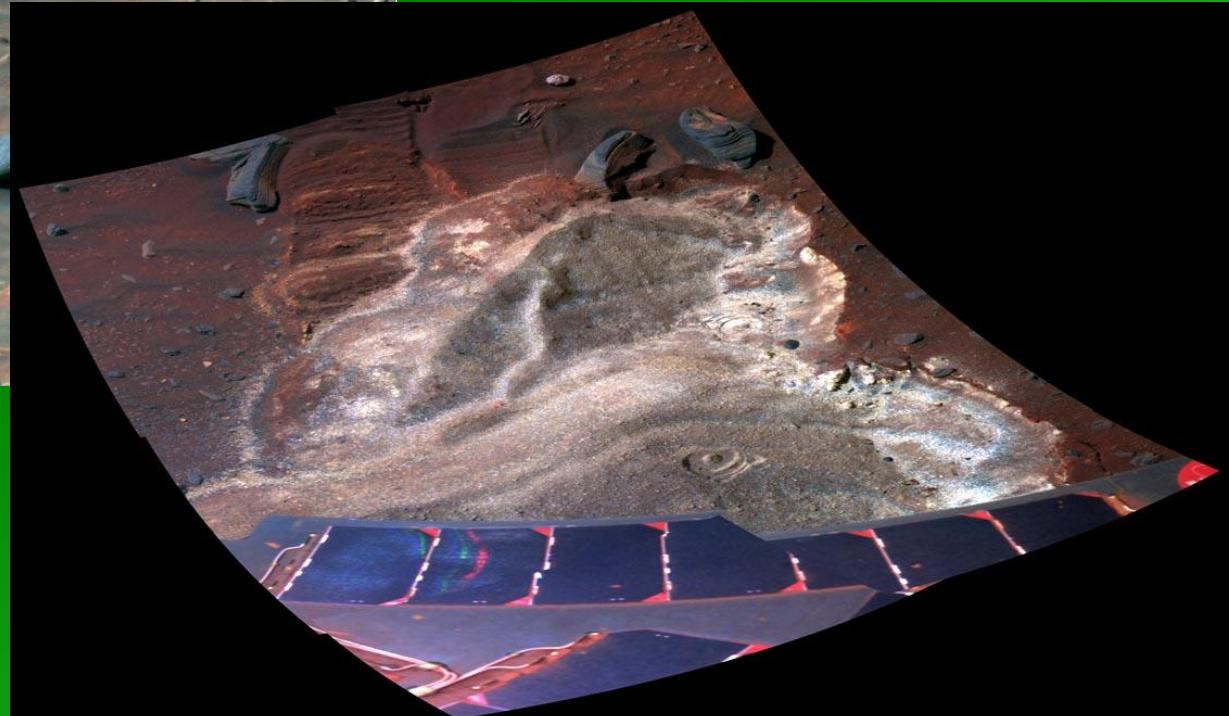
A Spirited Road



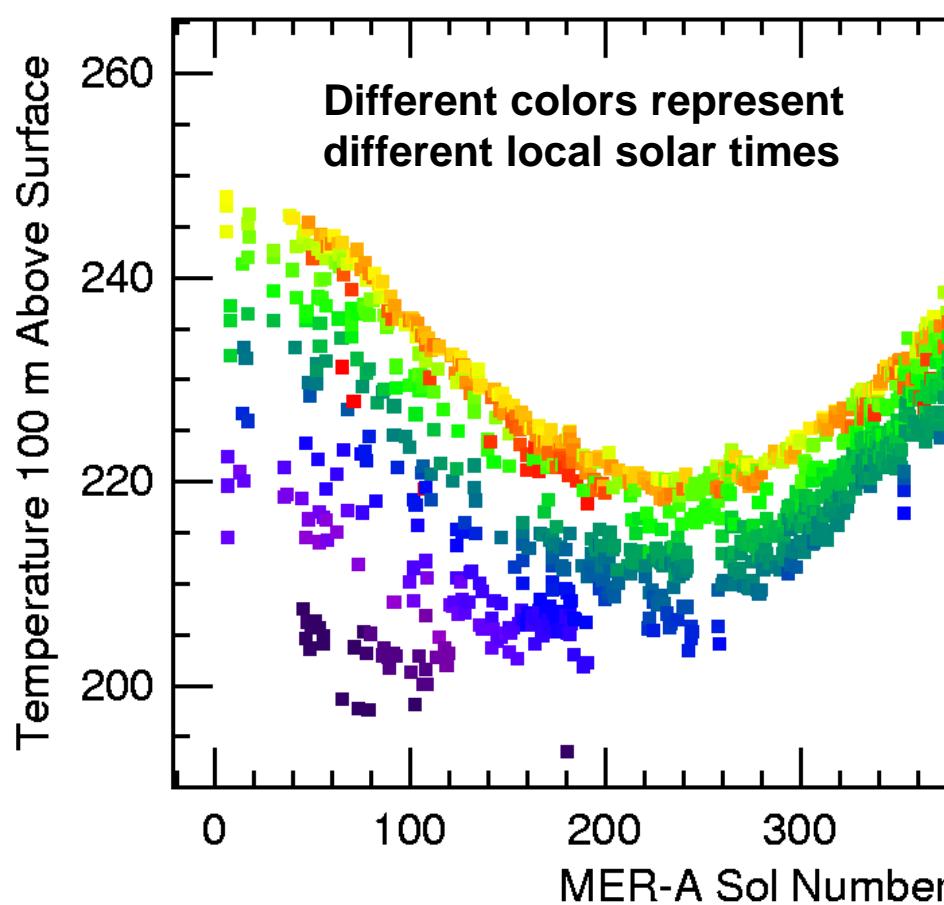


Subsurface salty soils

Mainly, Hydrous
Sulfates (Mg, Ca, Fe)

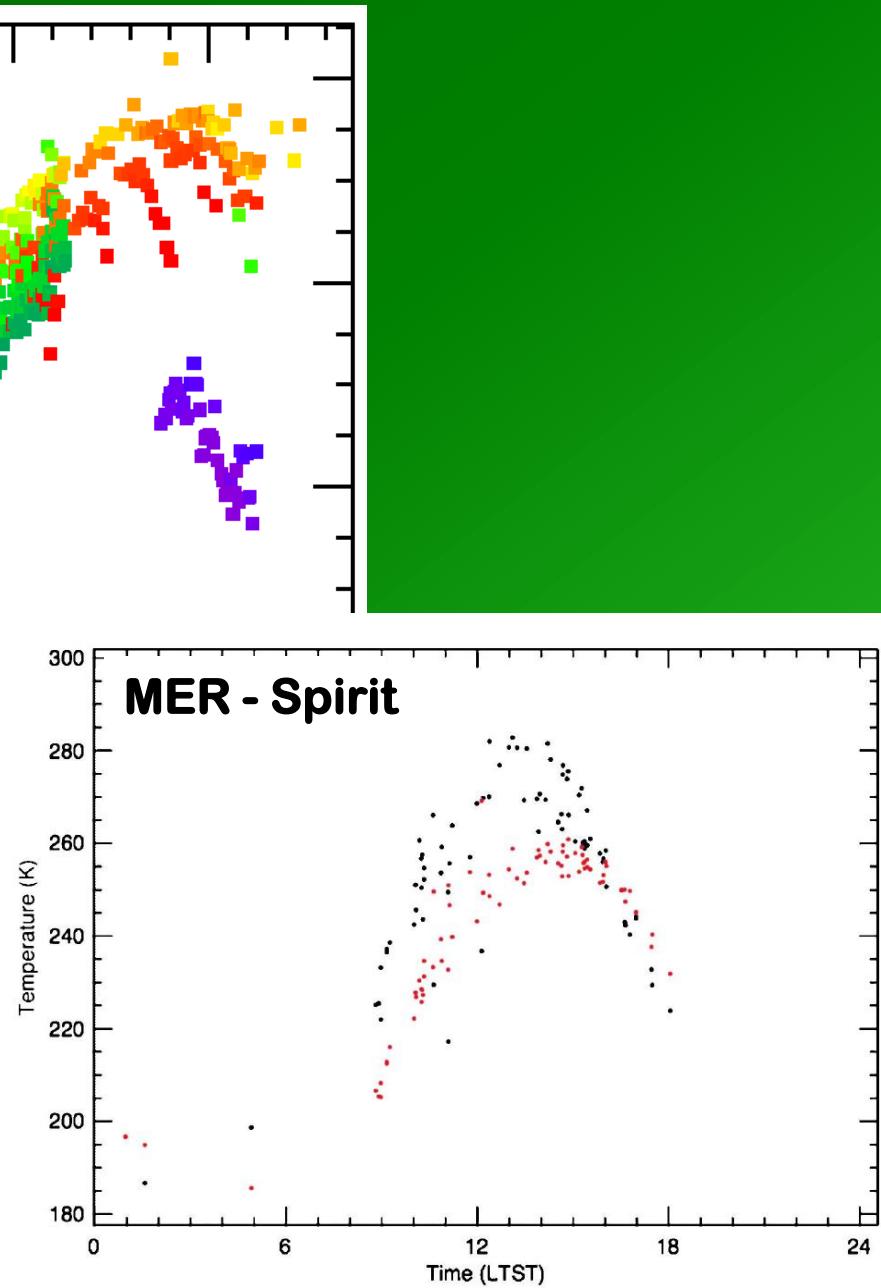


Atmospheric (near ground) Temperature Variation



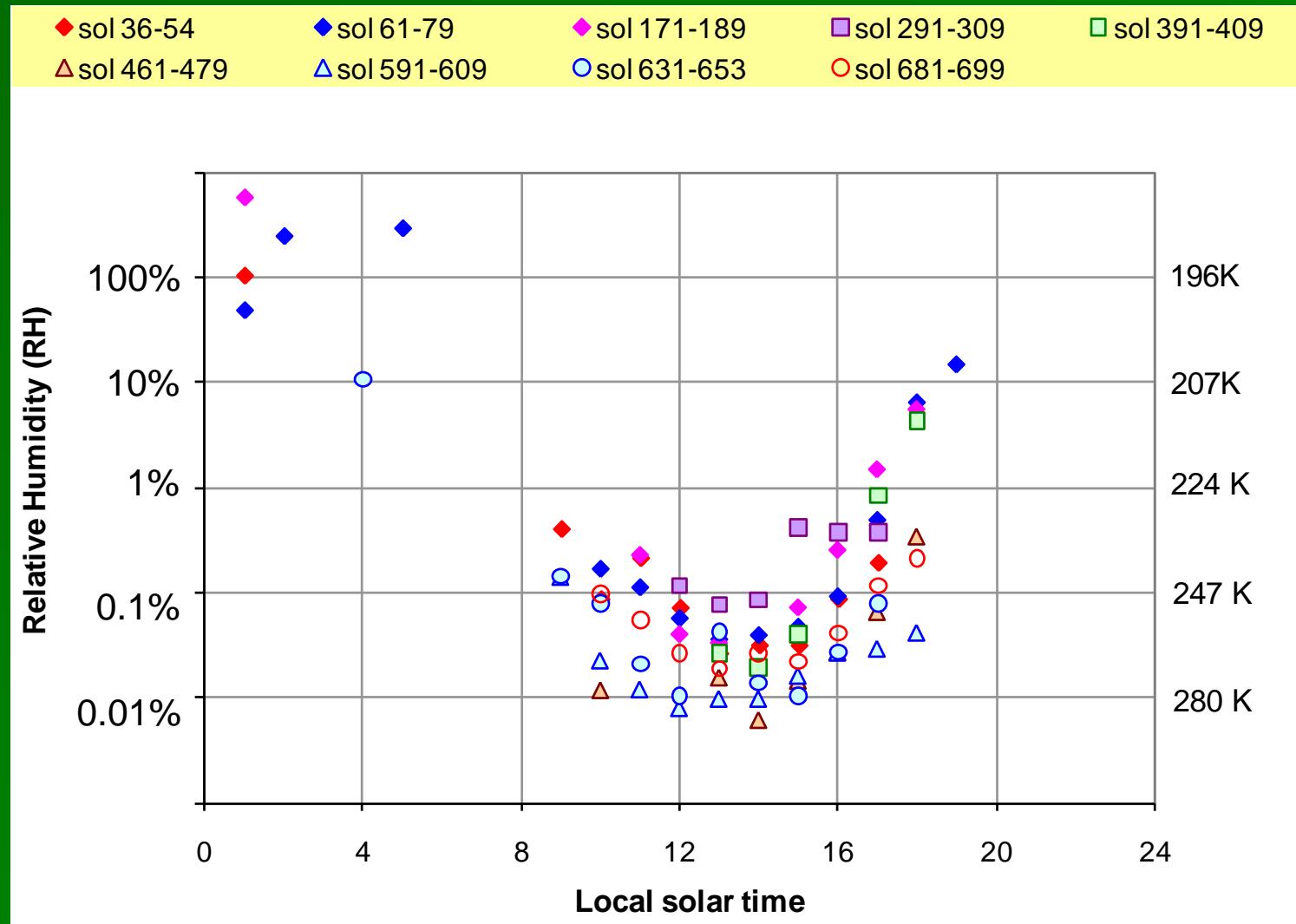
Smith et al., 2005

Spanovich et al., 2006

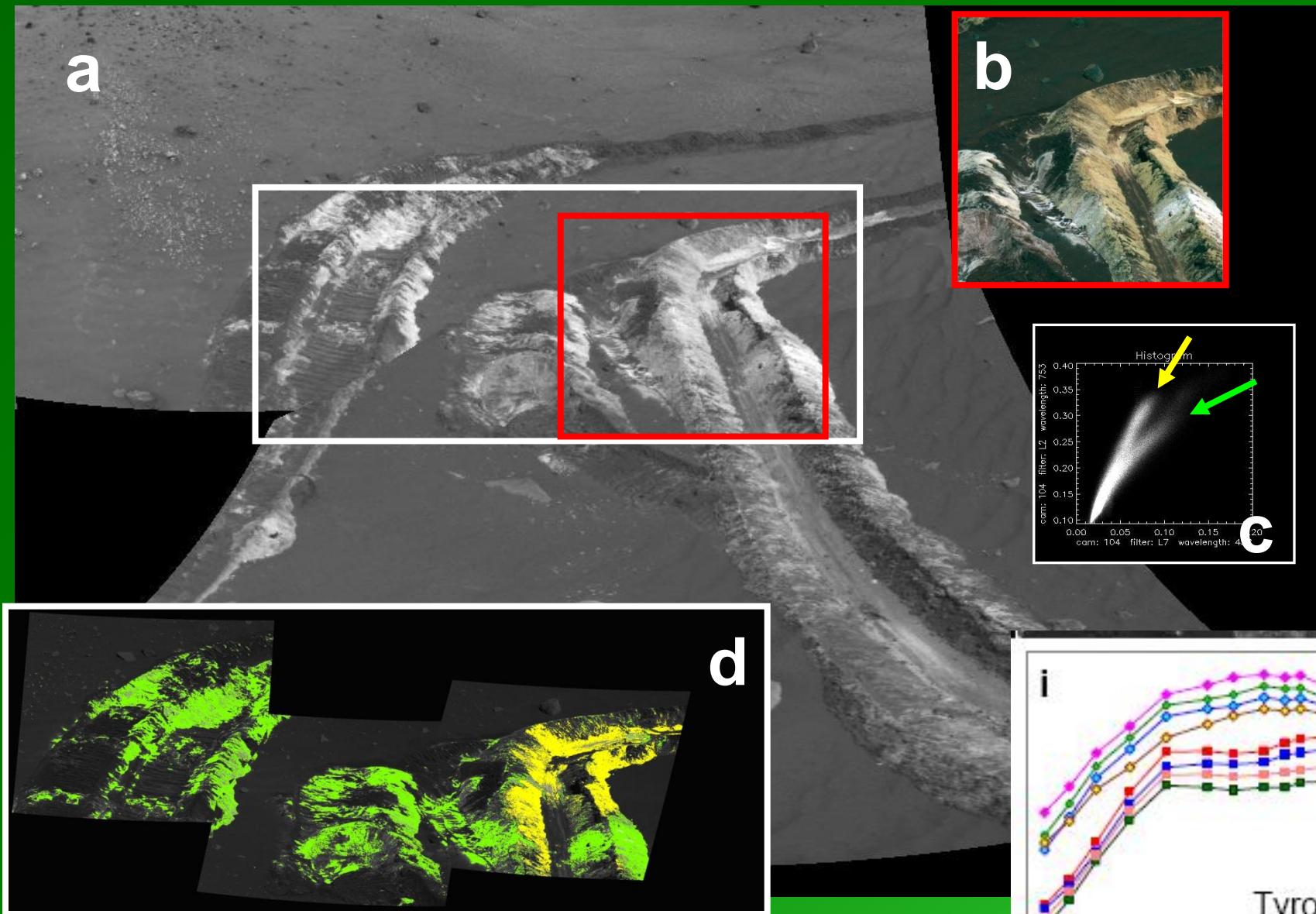


Calculated Variation of Relative Humidity

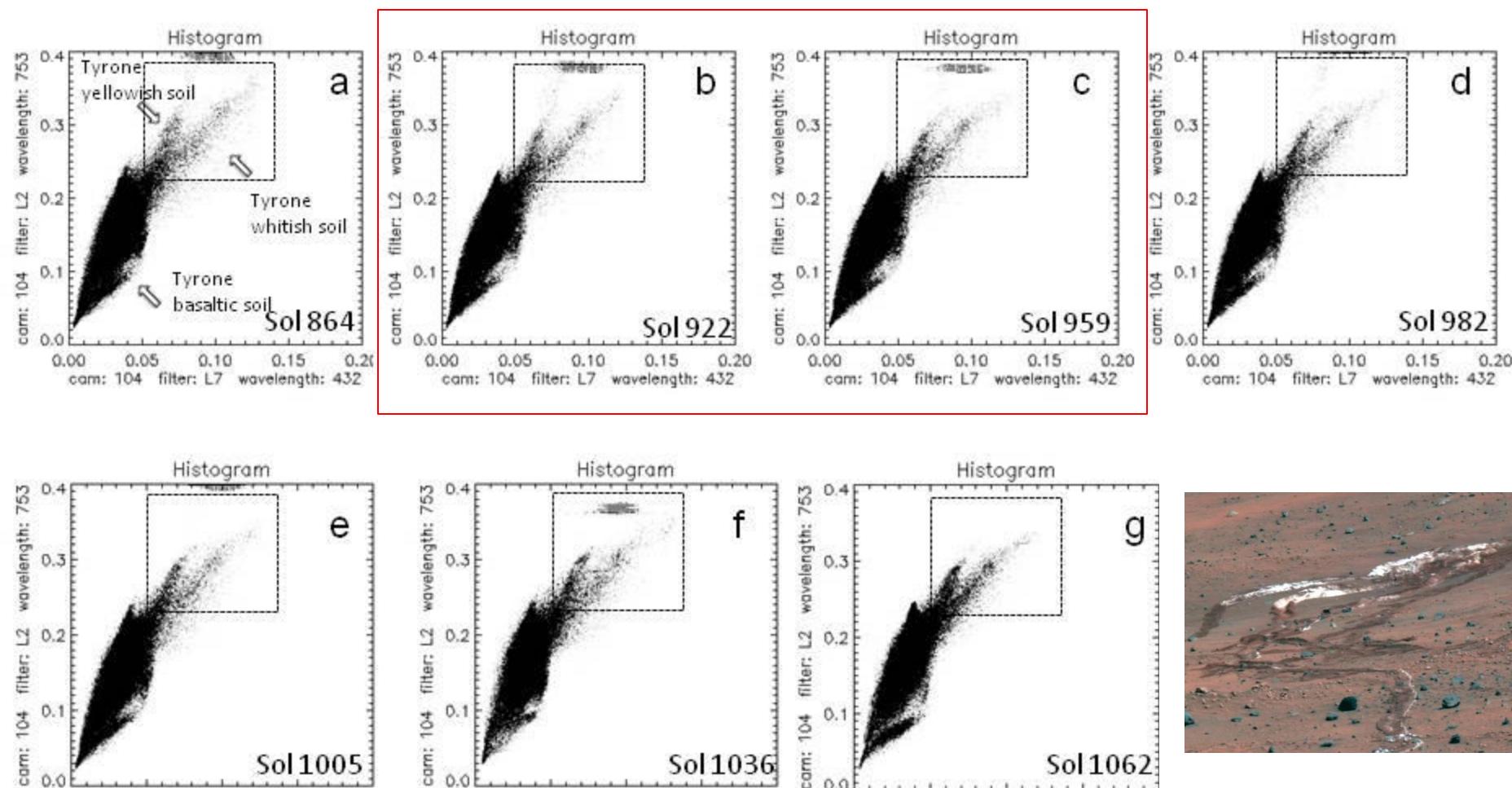
H_2O vapor pressure (ave): 10^{-2} kg/m^2



Two distinct layers of salty soils at Tyrone



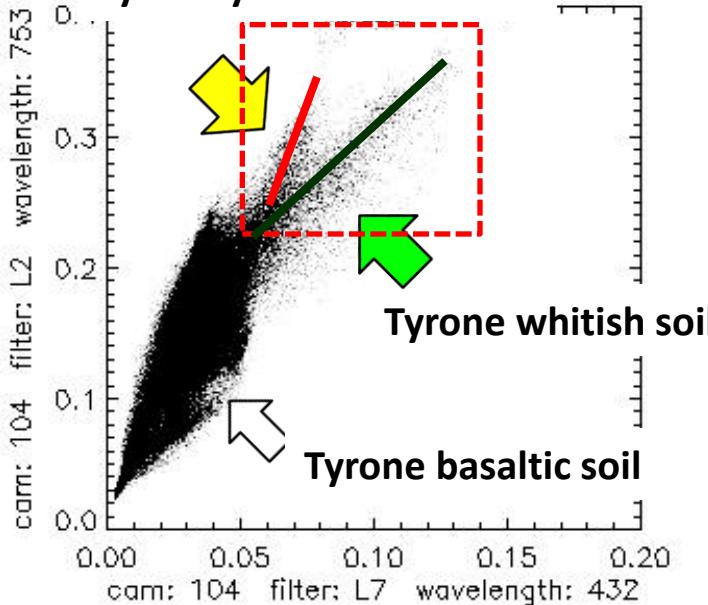
Seven Pancam Observations



Spirit stayed at the **same location** for 198 sols,
7 measurements took at *almost same local solar time*,
w/very **small tau variation** during the most part of 198 sols.

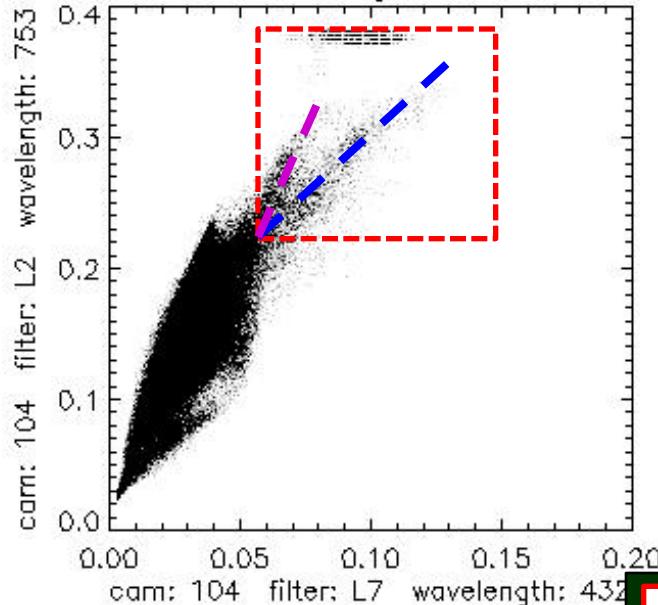
Property change of Fe-sulfate-rich Subsurface salty soil (R^*_{L2} vs. R^*_{L7})

Tyrone yellowish soil

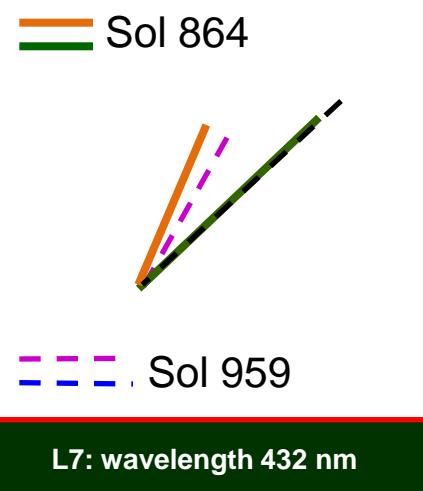


Sol 864

Histogram



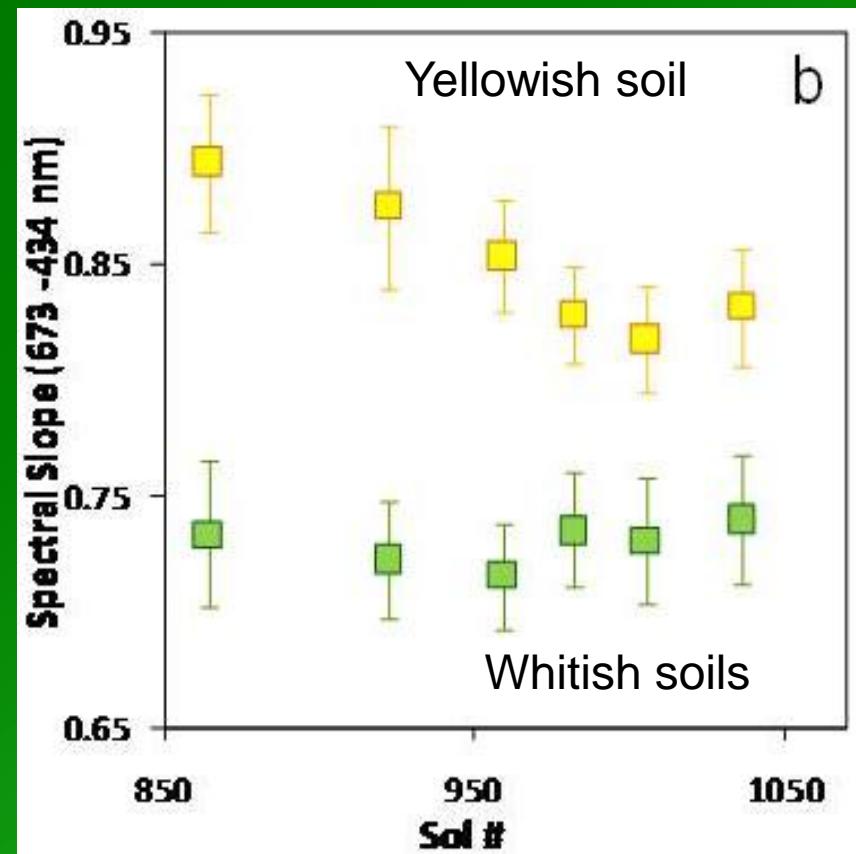
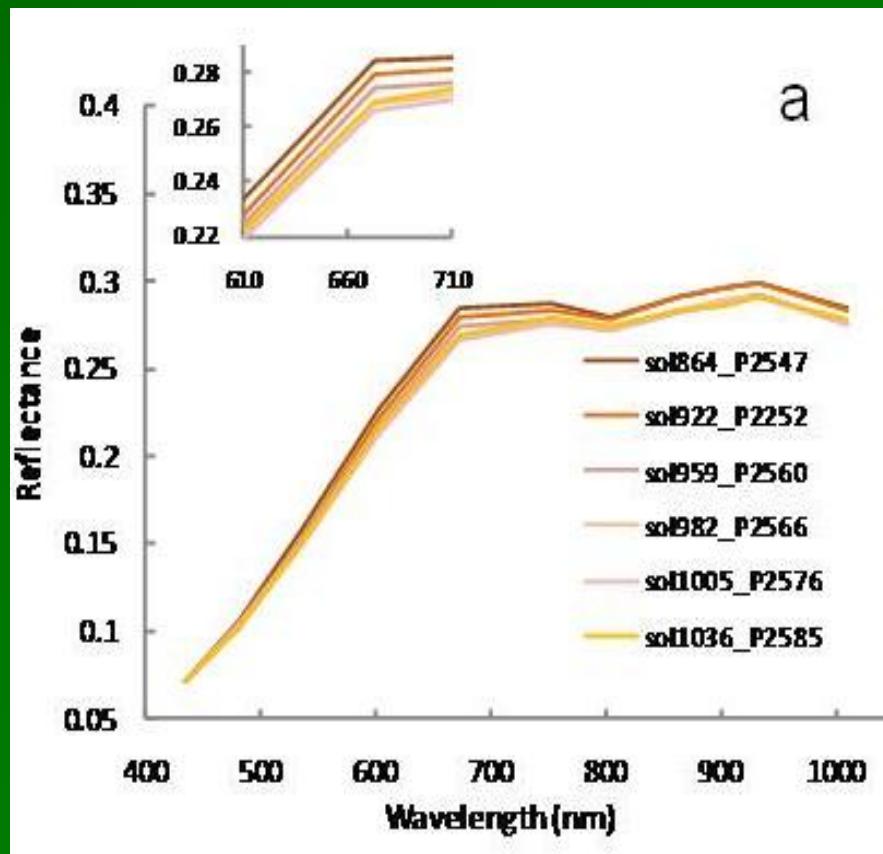
Sol 959



Change of spectral
slope of deep soils

L7: wavelength 432 nm

Selection of AOI → salty soils from deeper depth show 434 -753 nm spectral slope reduction.



Johnson et al., 2007 -- One of the major component of yellowish soils is ferricopiapite.

Lab observation #1: Phase transition Pathway

Ferricopiapite (21C), Wang et al., 2010

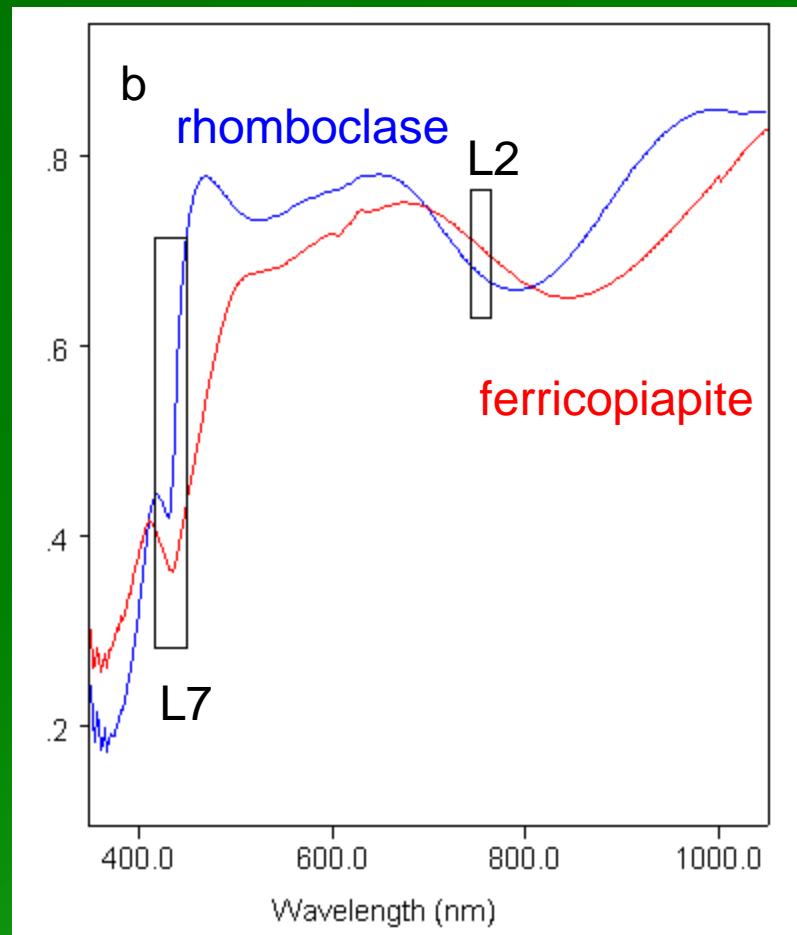
| RH | 6.60% | 11.30% | 33.00% | 54.10% | 58.80% | 69.70% | 75.40% | 85.00% | 94.40% | 100% |
|-------|-----------|------------|------------|-------------|------------|------------|--------------|--------------|--------------|--------------|
| hours | | | | | | | | | | |
| 2 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri |
| 8 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri |
| 20 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri |
| 48 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | ferri, deliq |
| 96 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | ferri, deliq | Deliq |
| 216 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | ferri, deliq | Deliq | Deliq |
| 384 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | ferri, deliq | Deliq | Deliq |
| 552 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | ferri, deliq | Deliq | Deliq |
| 720 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri, Delic | Deliq | Deliq | Deliq |
| 1080 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri, Delic | Deliq | Deliq | Deliq |
| 1992 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri, Delic | Deliq | Deliq | Deliq |
| 3840 | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri | Ferri, Delic | Deliq | Deliq | Deliq |
| 8664 | Ferri +Am | Ferri | Ferri+Rhor | Ferri | Ferri | Ferri+Rhor | Ferri, Delic | Deliq | Deliq | Deliq |
| 10488 | UK#9 | UK#9 | Ferri+Rhor | Ferri, para | Ferri | Ferri | Ferri, Delic | Deliq | Deliq | Deliq |
| 14520 | Ferri | Ferri | Ferri | Ferri, para | Ferri | UK#20 | Ferri, Delic | Deliq | Deliq | Deliq |
| 17688 | Ferri | Ferri+Rhor | Ferri | Ferri, para | Ferri, 7w | Ferri | Deliq | Deliq | Deliq | Deliq |
| 24912 | Ferri | Ferri | Rhom, Fer | P9w, Ferri | P9w, Ferri | Ferri | Deliq | Deliq | Deliq | Deliq |

Ferricopiapite dehydrates through two pathways:
pathway #1 → 9w, 7w, rhomboclase (4w);
pathway #2 → Amorphous species (5-11w, 14-19w).

Ferricopiaite → Rhomboclase

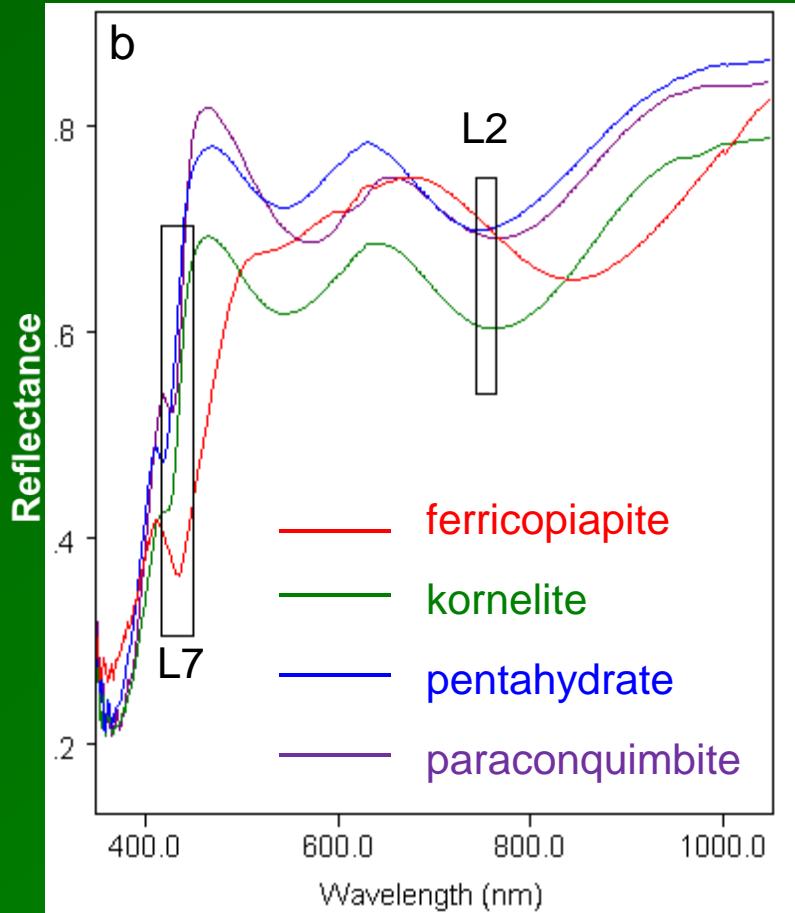


When the last drop of liquid dried

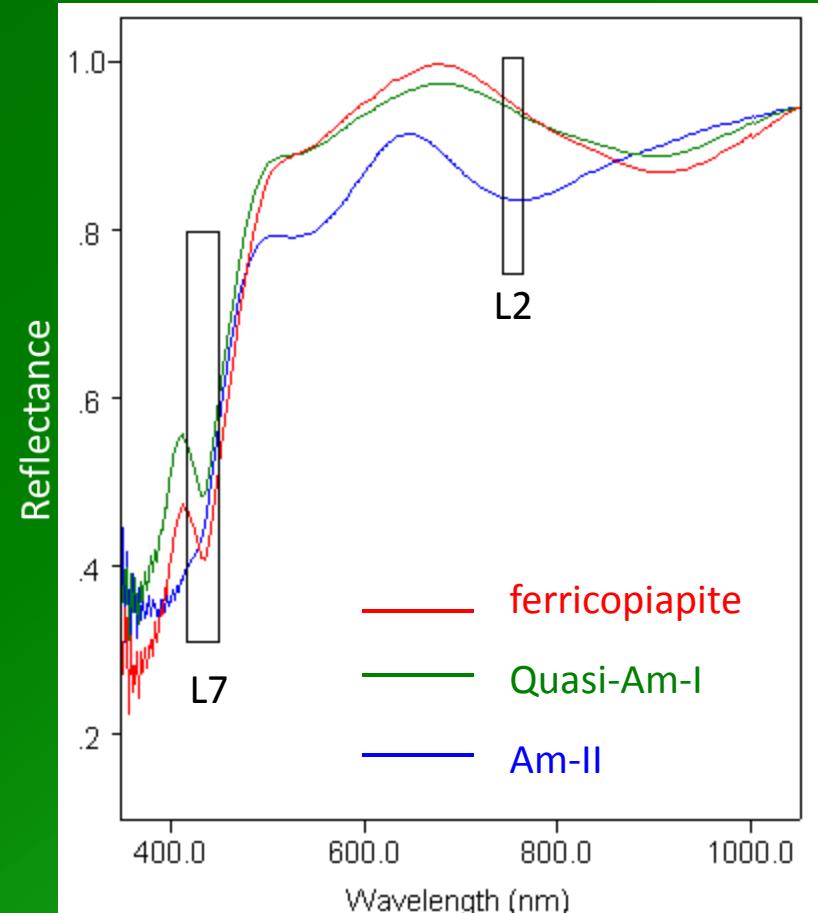


It will cause the spectral slope reduction from 434 nm to 753 nm (*Wang & Ling, 2011*)

pathway #1 → 9w, 7w, Rhom

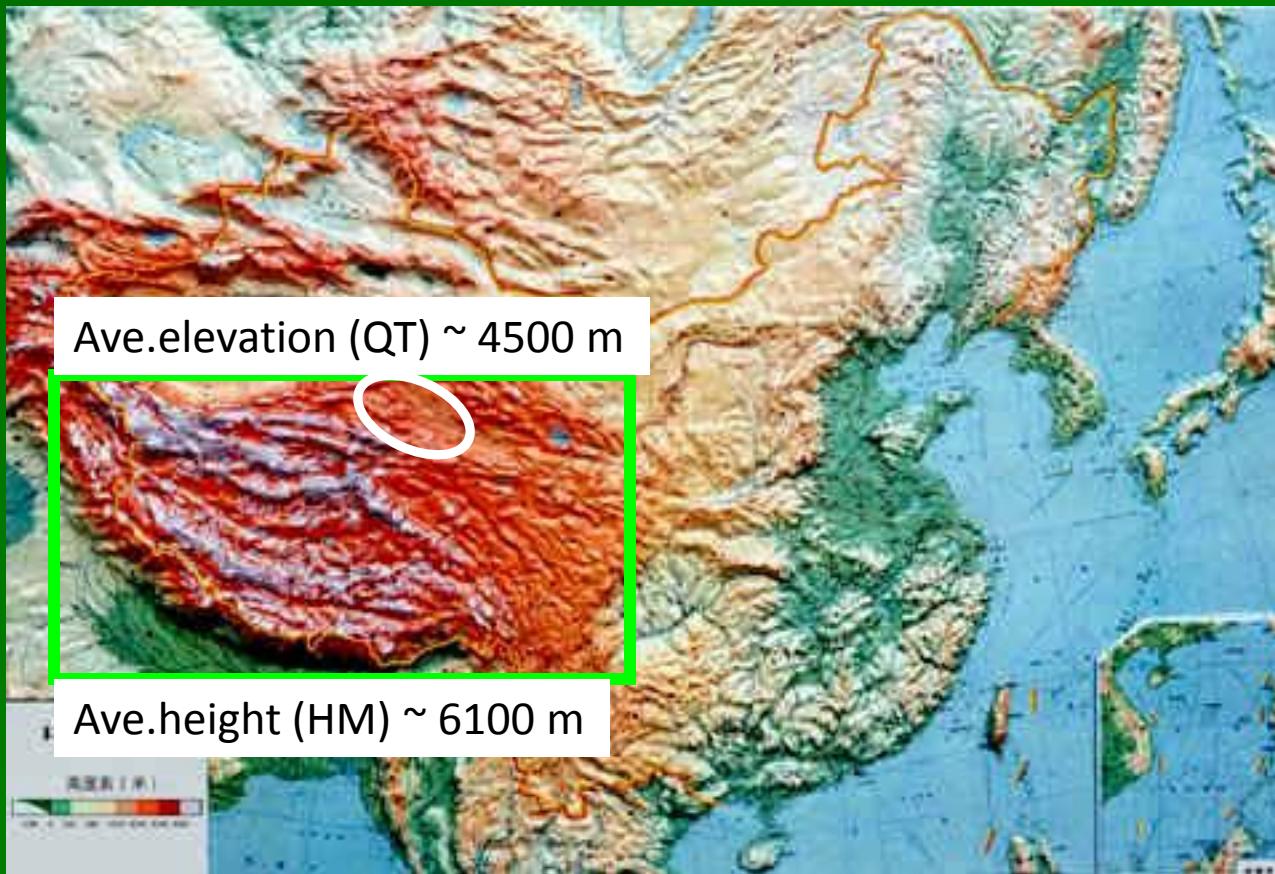


pathway #2 – Am (I, II)



The spectral change of Tyrone subsurface soil is consistent with dehydration of ferricopiapite

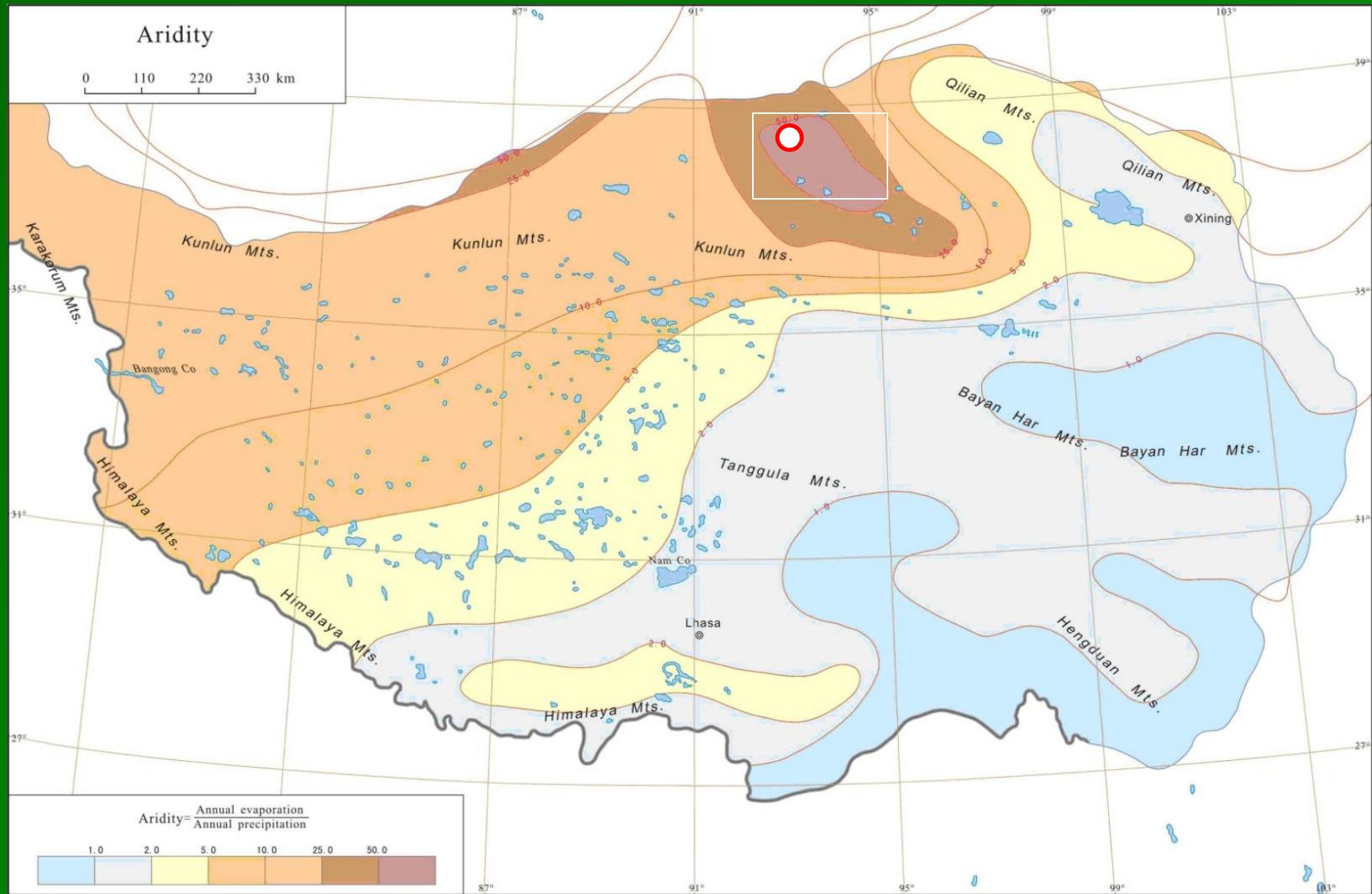
Terrestrial Analog Site Study (Tibet)



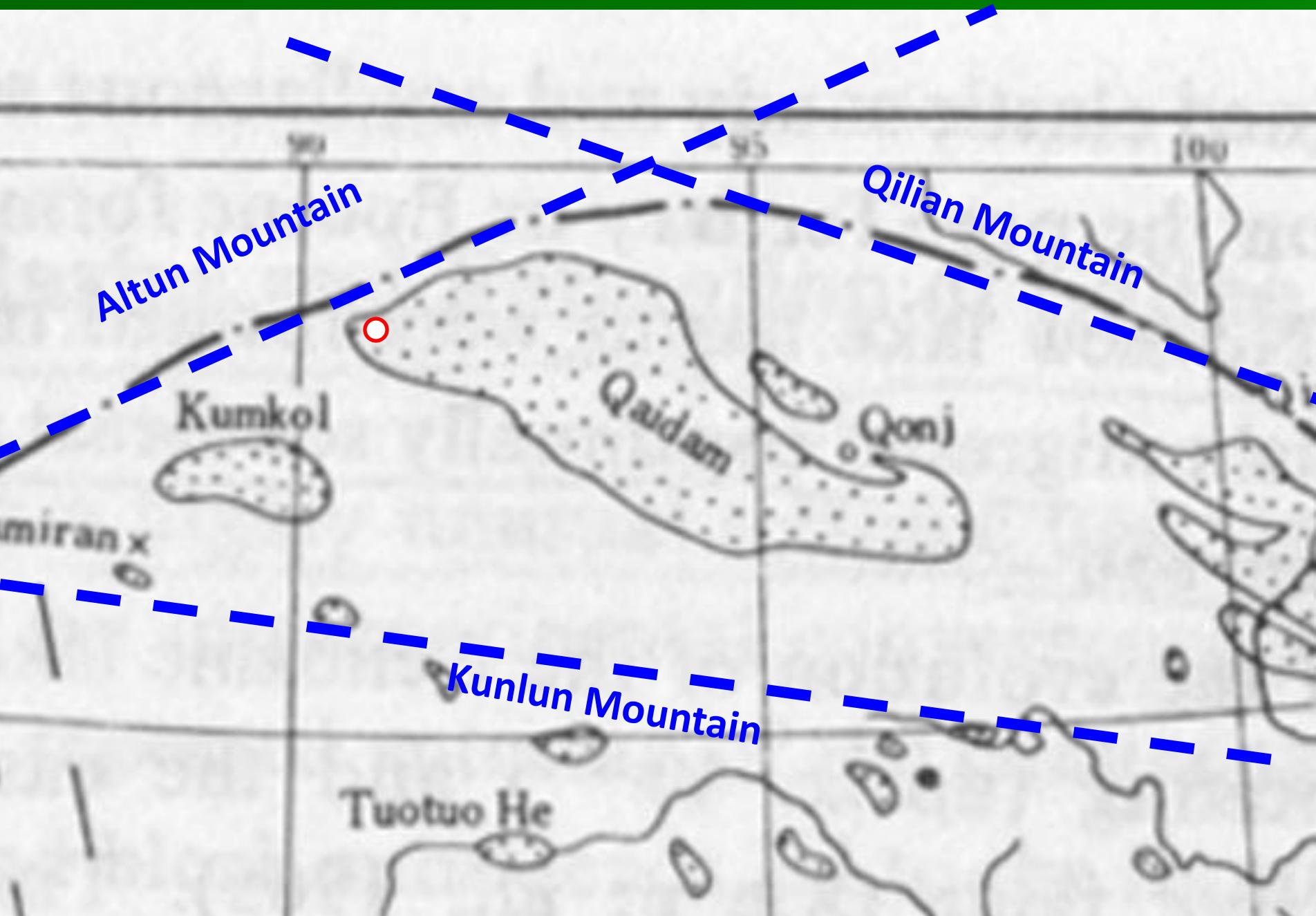
Climate conditions

- low P, low T, large ΔT , high UV, & **hyperarid**

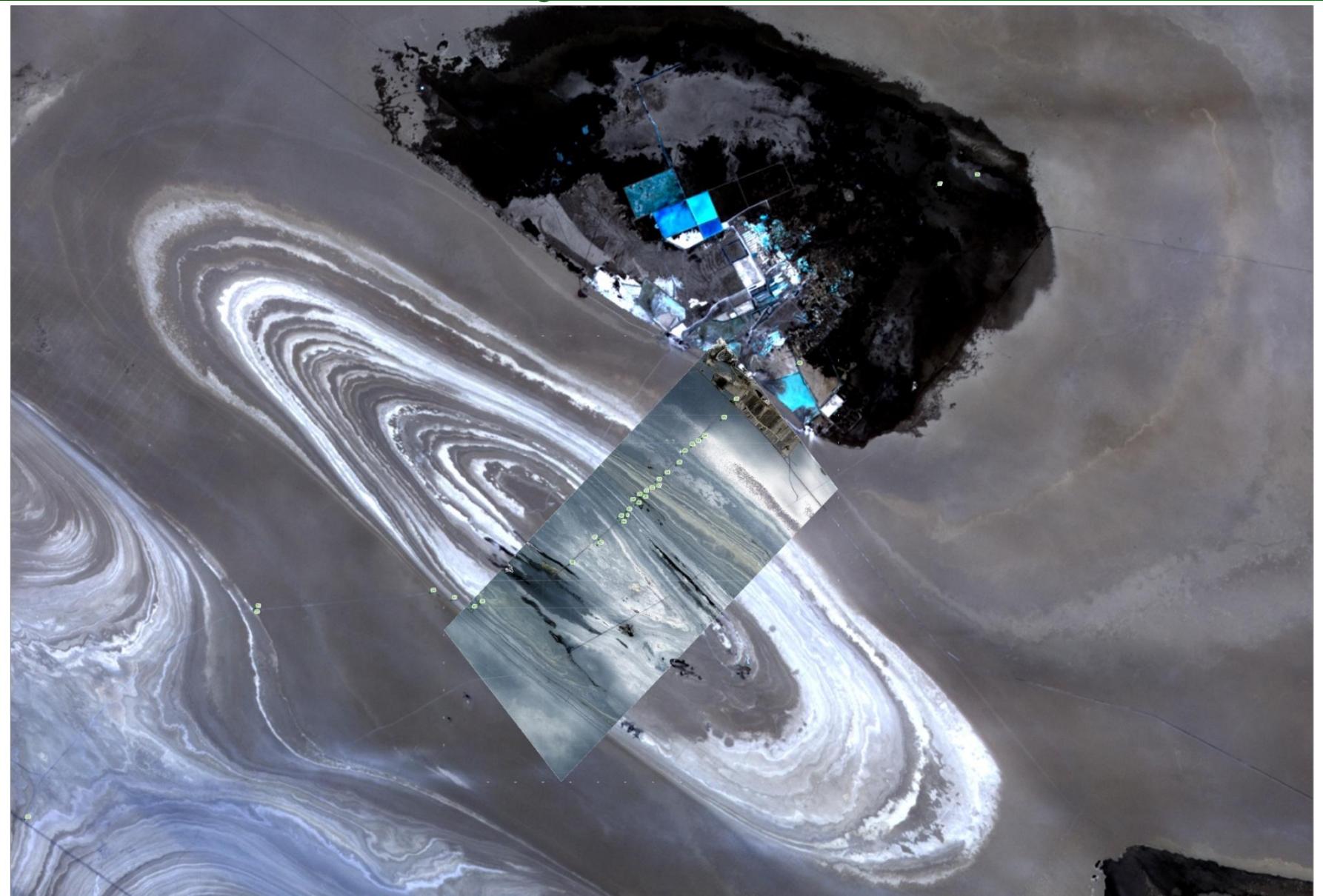
Aridity map of QT : DLT has the highest aridity ($AI < 0.01$)



Qaidam paleolake (Pliocene-early Pleistocene)



Satellite photo of Da Langtan (DLT) area



Legend

● Mtn. GPS Points at Da Langtan/Yao Lian She

1,600 800 0 1,600 3,200 4,800
Meters



General View of the area



Satellite photos of DLT w/sampling sites

R=15 m

05

01

03 & 04

02

06

R= 60 cm



1,600 800 0 1,600 3,200 4,800 Meters

Legend

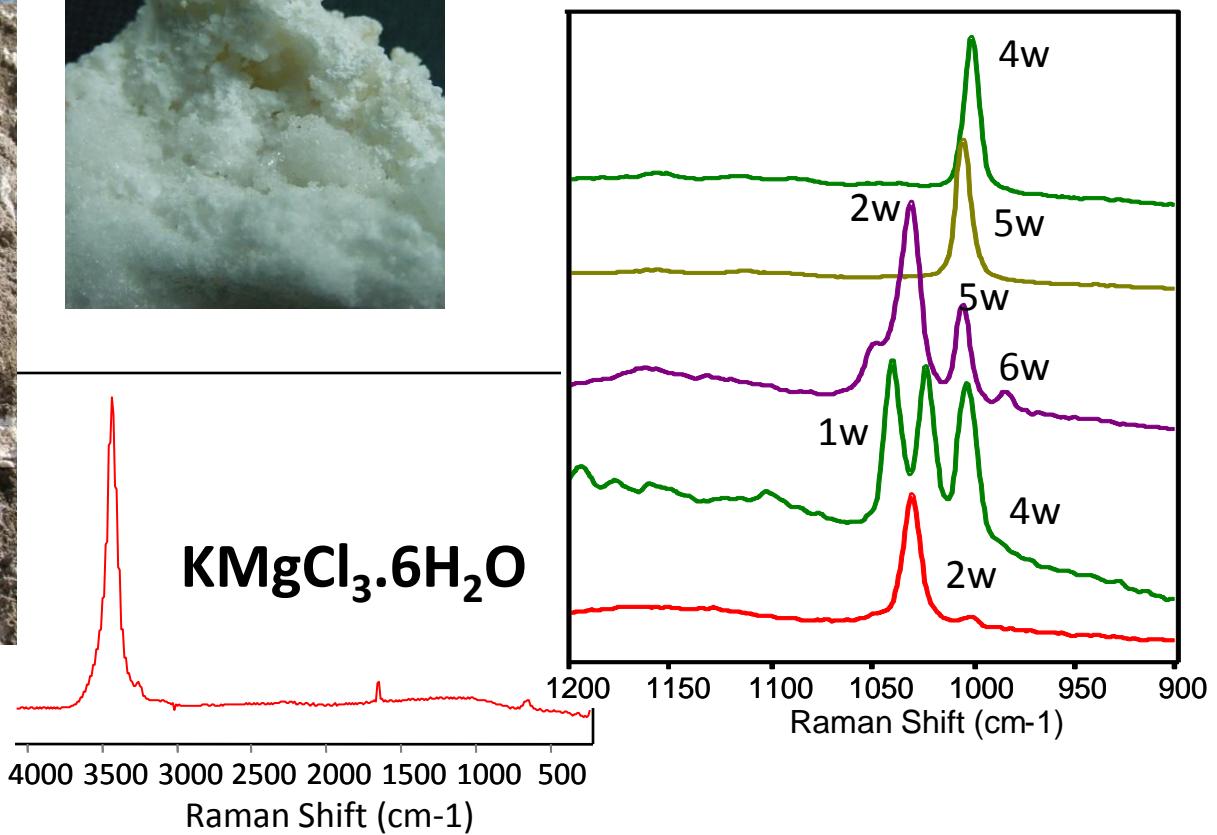
Sampling through a cross section



Subsurface salts w/high degree of hydration



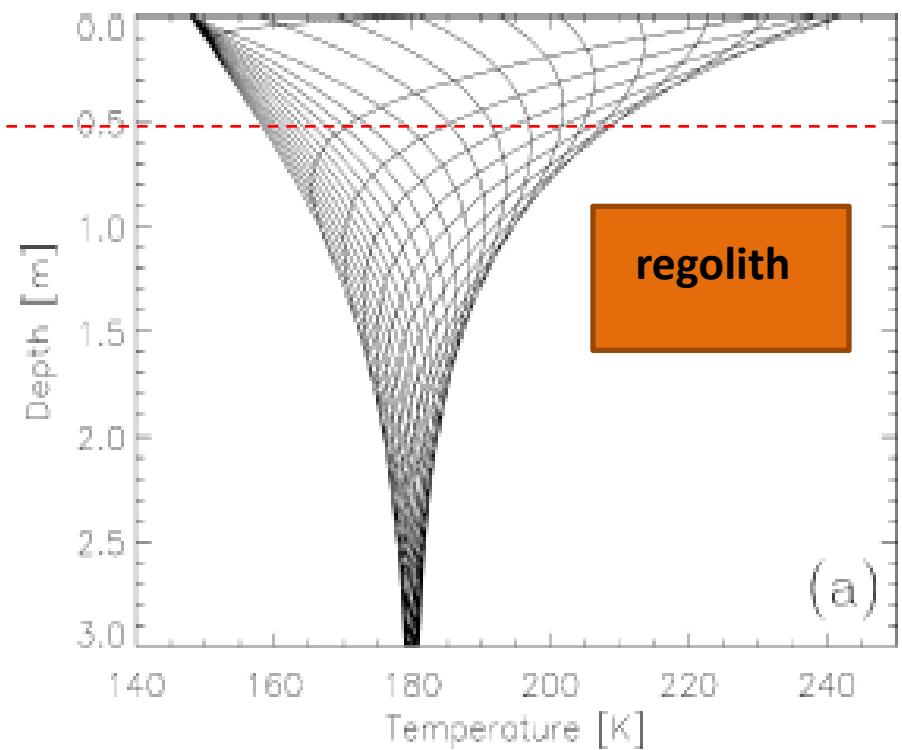
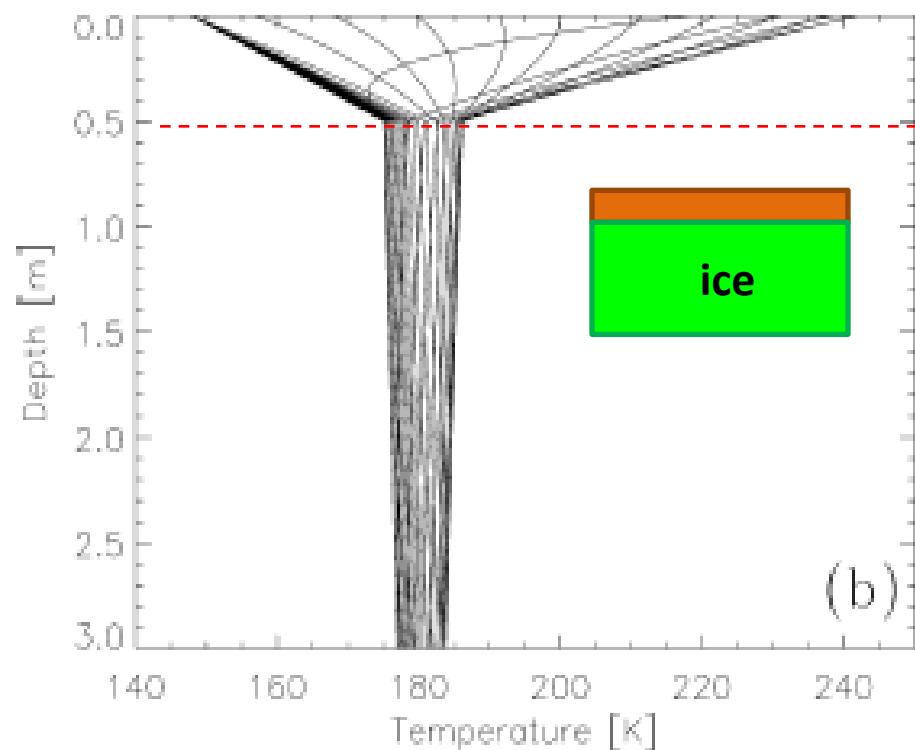
$\text{MgSO}_4 \cdot (4\text{-}6) \text{ H}_2\text{O}$



Highly hydrated salts are preserved in the *subsurface* of a hyperarid region.

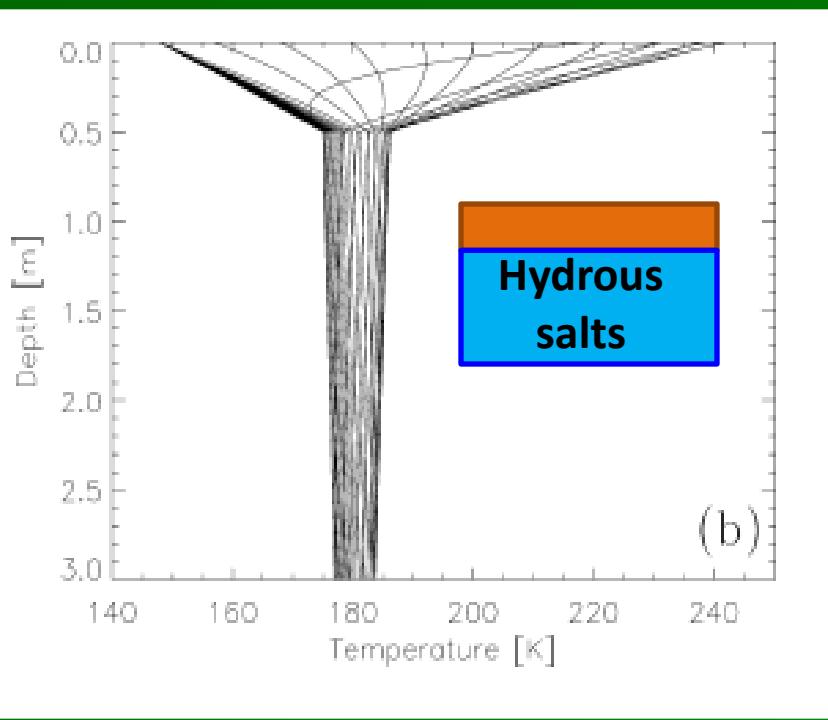
A thermal model by Mellon et al., 2004

T profile in regolith



Ice -rich subsurface: low T_{ave} & small ΔT

Extrapolation of Mellon's model



Assumption -- Hydrous salts have higher thermal inertia than surface regolith.



Lower T_{ave} , smaller ΔT



Higher RH, smaller ΔRH

Atmospheric conditions at the surface have less influence in salt-rich subsurface ;
Large quantity of hydrous salts will buffer the RH of the environment

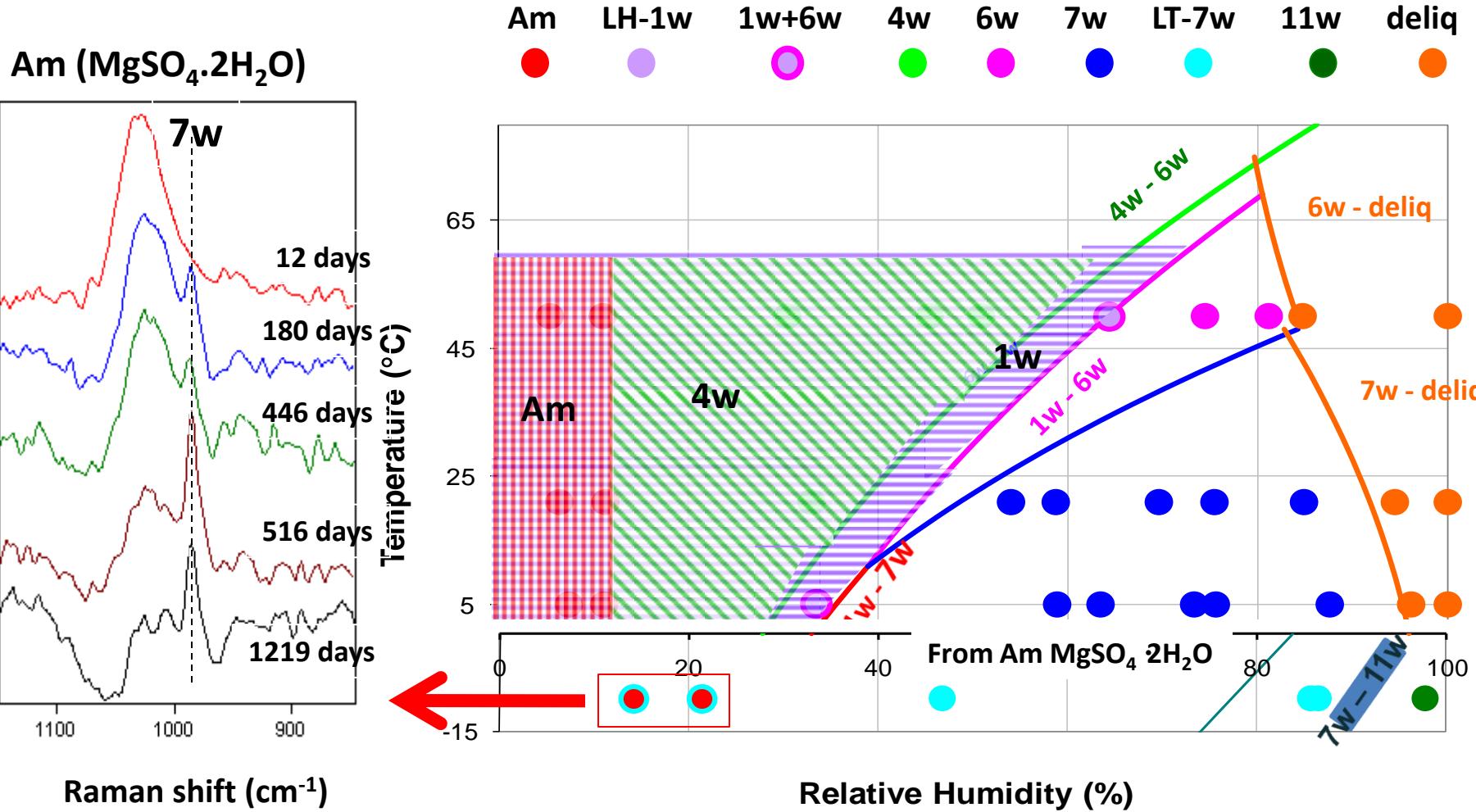
Laboratory observation #2

At -10 °C, a space filled w/MgSO₄.7H₂O
RH ~ 96 – 97%

At -10 °C, a space filled w/ferricopiaite
RH ~ 75 -79%

W/enough quantity,
Highly hydrated salts can keep a high
RH within subsurface

Laboratory observation #3



Large stability fields of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ @ low T

Mars -- salt-rich subsurface at equatorial region →
Low T, high RH ;

Earth -- subsurface salty layer in a **hyperarid**
region → **High RH;**

Thermal model -- A salt-rich subsurface would be
less influenced by atmospheric conditions;

Lab – highly hydrated salts can **buffer a high RH**
environment at low T;

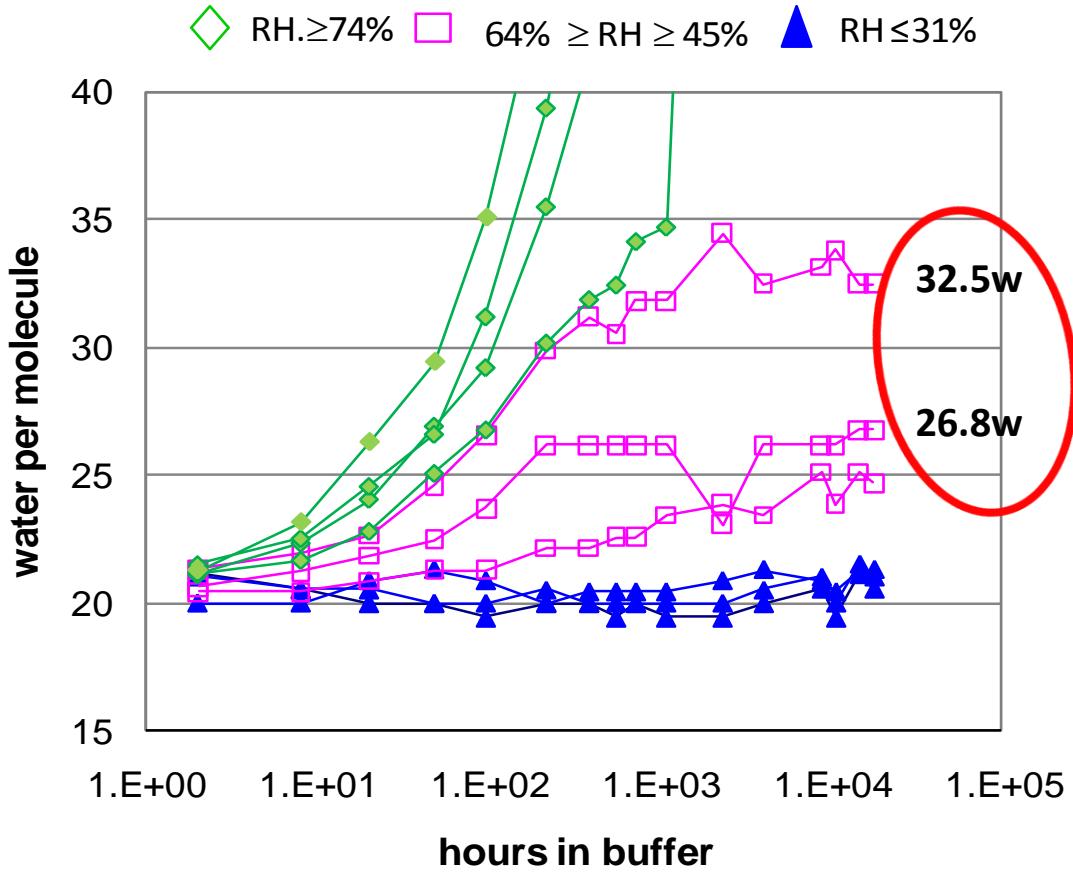
Lab – highly hydrated salts have **large stability**
field (RH) at low T.



What about Liquid water...

Laboratory observation #4

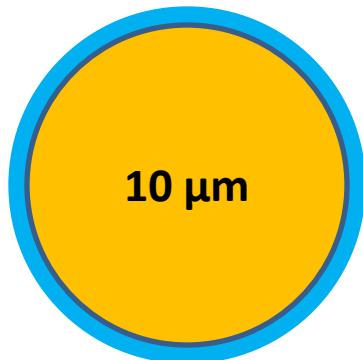
Figure 4. $\text{Fe}_{4.67}(\text{SO}_4)_{6}(\text{OH})_{2.20}\text{H}_2\text{O}$ at 5°C



26.8w, 32.5w suggest extra 6.8w – 12.5w at grain surfaces

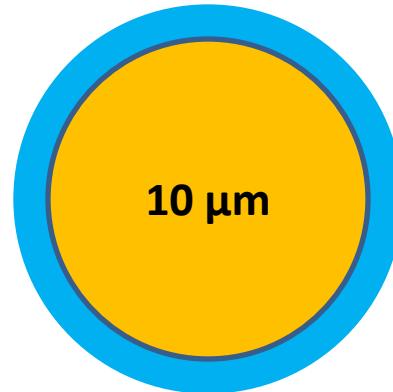
The meaning of extra **6.8w – 12.5w** per ferricopiaite molecule

6.8w/molecule



**1137 layers
of H₂O
(0.35 μm)**

12.5w/molecule



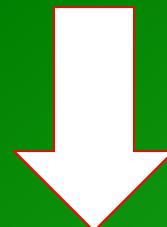
**2090 layers
of H₂O
(0.65 μm)**

**Thin film of liquid H₂O has formed at
the grain surfaces of ferricopiaite**

Low T, high RH, H₂O film

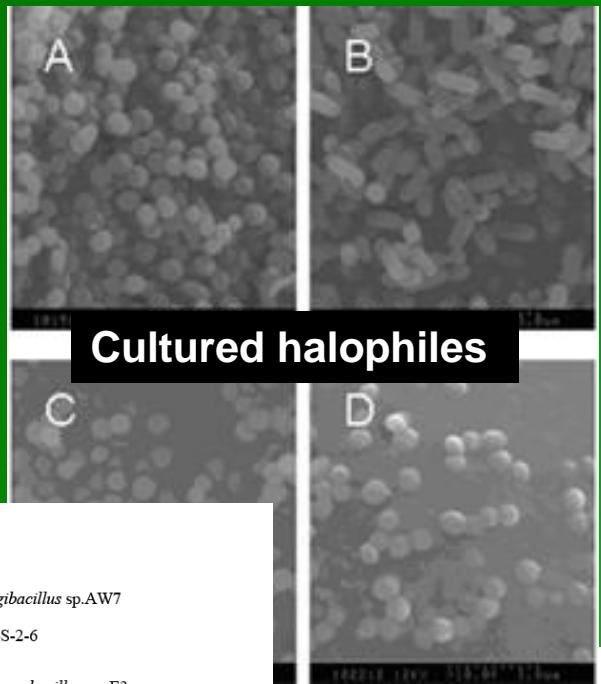
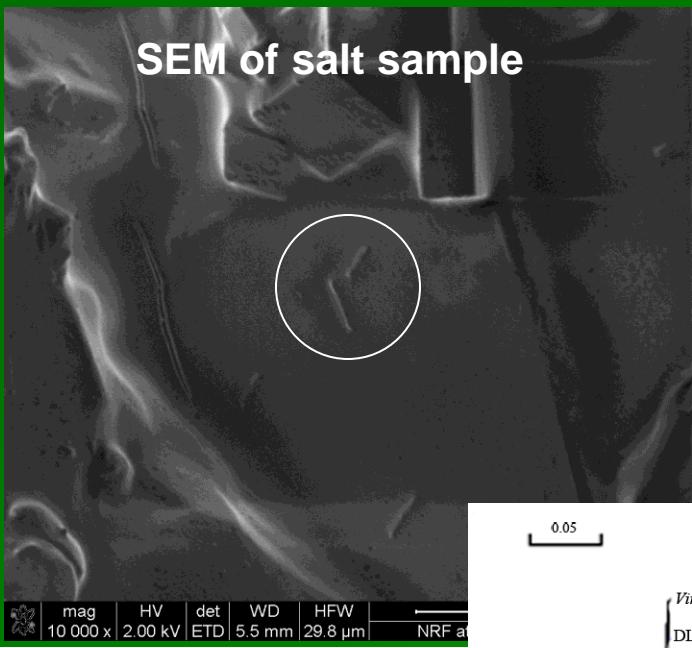


Habitable ?

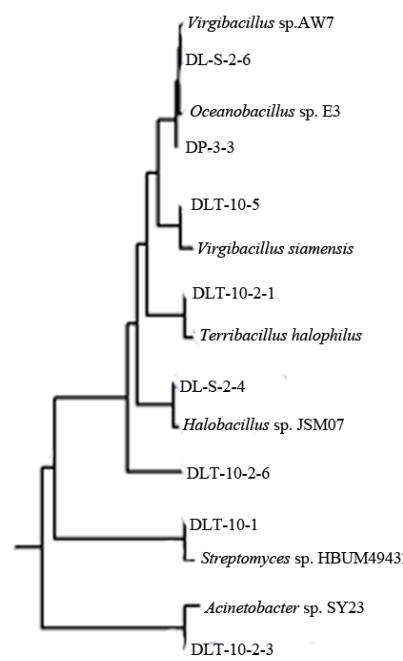
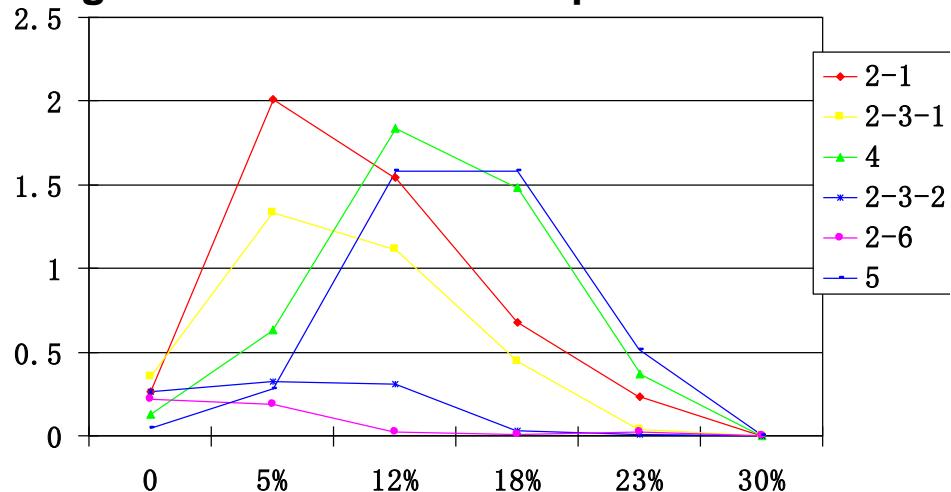


Inhabited ?

Inhabited at QT -- halophiles

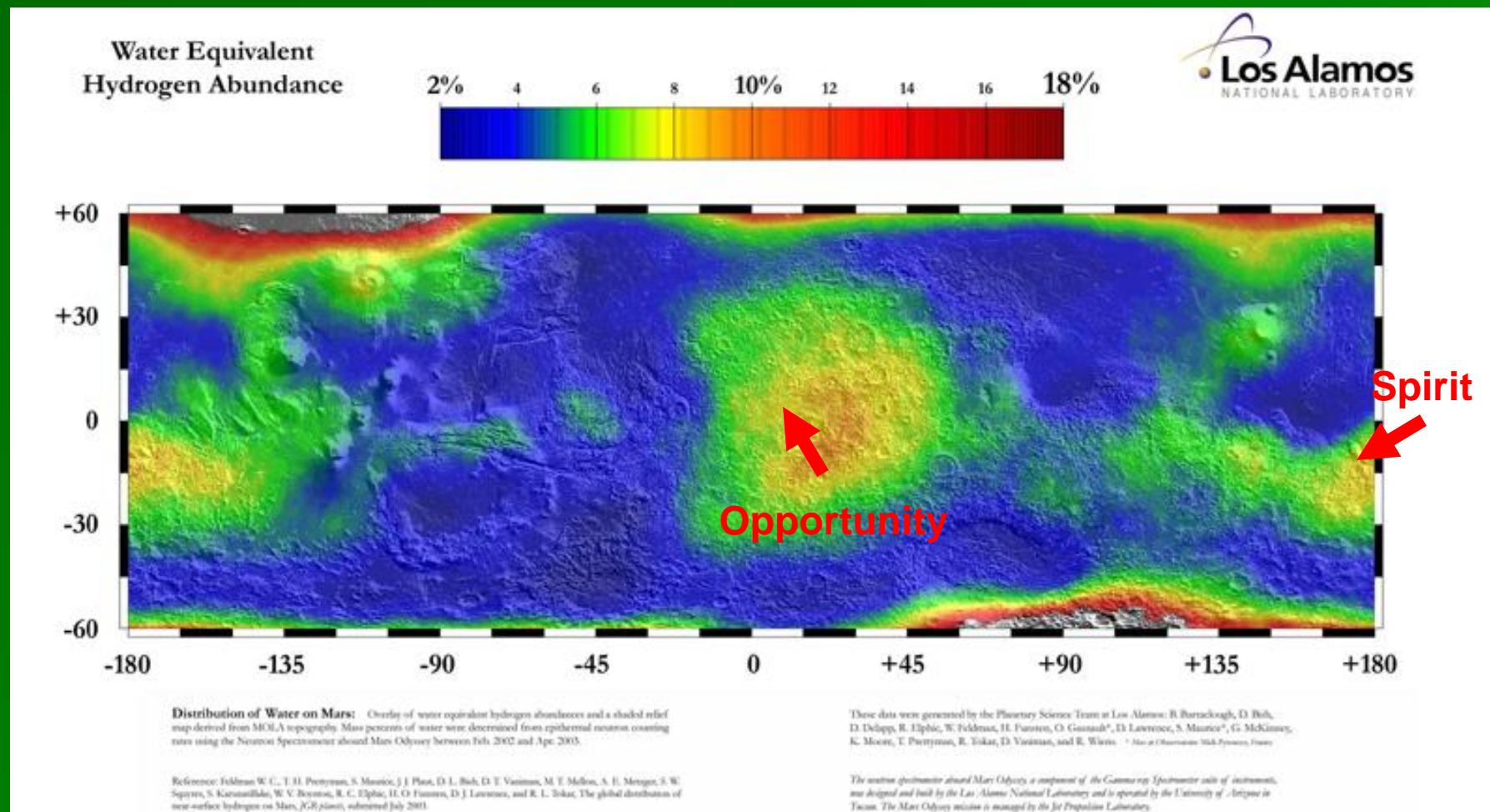


The growth curve of six halophile strains



16S rRNA
gene sequences
for eight strains
halophiles

Subsurface in equatorial regions on Mars ?



Habitable?

inhabited ?